

(1) (a) (1) \rightarrow (2) adiabatic compression, liquid

(2) \rightarrow (3) expansion, approximately isobaric, phase transition $l \rightarrow v$

4 (3) \rightarrow (4) adiabatic expansion, vapour + initial phase transition $v \rightarrow l$

(4) \rightarrow (1) isothermal "compression", \nparallel full phase transition to liquid

$$\text{1) principle: } 2 \eta = \frac{\text{Work done}}{\text{heat invested}} = \frac{W_H - W_P}{Q_H} = \frac{Q_H - Q_L}{Q_H} = 1 - \frac{Q_L}{Q_H}$$

$$Q_L = (S_4 - S_1) T_L$$

$$2 Q_H = \int_2^3 \frac{S dt}{T ds} \rightarrow \eta = 1 - \frac{(S_4 - S_1) T_L}{\int_2^3 \frac{S dt}{T ds} T ds}$$

(c) approximation $\int_2^3 \frac{S dt}{T ds} \approx (S_3 - S_2) T_H$ Path of phase transition is much longer
 3 than the other two. And even better, we overestimate the first part this way, but
 underestimate the last.

(d) 2 These trajectories are not adiabatic anymore

$$3 \quad \eta = 1 - \frac{T_L}{T_H} \quad T_L = 115^\circ C = 388 K$$

$$\hookrightarrow \eta = 50.4\%$$

from figure 3 we read $T_H = 400^\circ C = 673 K$, $T_L = 130^\circ C = 403 K$

$$\eta = 40\%$$

$$(2) (a) \quad a_a = \frac{(100 - 24)}{340} = 0.2235 \dots$$

3

$$t.a = \frac{185}{340} = 0.5441 \dots$$

$$a_s < \frac{24}{185} = 0.1297 \dots$$

- (a) sensible heat : actual heat transported upwards by "warm" air
 9 latent heat: the heat of condensation transported up in the form of water vapours
 contained in rising air (which can potentially condensate) $C = 104 \text{ W/m}^2$

(c) 1 $10 - 25 \mu\text{m}$

(d) 1 $t_a^1 = 0.7 \rightarrow \text{atmospheric window} = 39.8 \text{ W/m}^2$

(e) 2 $\sigma T_s^4 = 398 \rightarrow T_s = 289.45 \text{ K}$

$$(f) \quad \begin{aligned} i \quad \sigma T_a^4 &= 342 & \rightarrow T_a &= 278.68 \text{ K} \\ ii \quad \sigma T_a^4 &= (239 - \underline{40}) = 199 & T_a &< 243.398 \text{ K} \end{aligned}$$

(g) temperature at low and high altitudes, respectively.

3 The low altitude air represents the air from which thermal infrared makes it back to the surface, ~~the~~ whereas only radiation from high up in the atmosphere finally makes it to outer space.

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(3)

3

- (a) This reaction is the "best accessible", that is the lowest energy needed 2 to get it going.

(b) ${}_{\text{2}}^{\text{3}}\text{T}$ is formed through a nuclear reaction with ${}_{\text{3}}^{\text{6}}\text{Li}$ and neutrons

(c)	${}_{\text{2}}^{\text{3}}\text{D}$	1.1 MeV/nucleon	$\times 2$	2.2
	${}_{\text{2}}^{\text{3}}\text{T}$	2.8 MeV/nucleon	$\times 3$	8.4

$$\begin{array}{l} \text{4} \\ \text{4 He} \quad 7.1 \text{ MeV/nucleon} \quad 28.4 \text{ MeV} \\ \rightarrow n^* \quad - \\ \hline 17.8 \text{ MeV per reaction} \end{array}$$

- (d) The plasma gets heated by this binding energy, heat transferred to the walls, 3 and through heat exchangers brought to an external steam turbine

$$\begin{array}{l} \text{4} \\ \text{3} \end{array} \text{(a)} \quad E = \frac{1}{2} J_m u^2 \quad J_m = g. A. \pi u. \Delta t \quad |$$

$$\rightarrow | \frac{E}{A. \Delta t} = \frac{1}{2} g u^3 |$$

NB Students performed very badly! Decided to compute marks with max points = 10 here and weight 5

- (b) The area from which wind energy is harvested is A_{IN} , not A . And A_{IN}/A depends on the velocity ratio u/u_{IN}

$$\text{2} \quad \text{Correct efficiency } \eta = \left(1 - \left(\frac{u_{out}}{u_{in}}\right)^2\right) \cdot \frac{A_{IN}}{A} = \left(1 - \left(\frac{u_{out}}{u_{in}}\right)^2\right) \cdot \frac{u}{u_{in}}$$

- (c) work = Force \times distance
 $=$ mass \times acceleration \times distance

$$1 = J_m \times \frac{u_{out} - u_{in}}{\Delta t} \times u \Delta t$$

and work = decrease of kinetic energy:

$$1 \frac{1}{2} \rho_m v_{IN}^2 - \frac{1}{2} \rho_m v_{OUT}^2 = -\rho_m \times \frac{U_{OUT} - U_{IN}}{\Delta t} \times u \Delta t$$

4)

$$\frac{1}{2} (v_{IN} - v_{OUT}) (v_{IN} + v_{OUT}) = -u (v_{OUT} - v_{IN})$$

$$1 u = \frac{v_{IN} + v_{OUT}}{2}$$

$$(d) \eta = \left(1 - \left(\frac{v_{OUT}}{v_{IN}}\right)^2\right) \cdot \frac{u}{v_{IN}} \quad \leftarrow v_{OUT} = 2u - v_{IN}$$

(*) if not found in (d) start with anything else

$$= \left(1 - \left(\frac{2u - v_{IN}}{v_{IN}}\right)^2\right) \frac{u}{v_{IN}}$$

work toward) an expression with u/v_{IN} (of the like)

$$= \left(1 - \left(\frac{2u}{v_{IN}} - 1\right)^2\right) \frac{u}{v_{IN}}$$

$$\downarrow = \left(\frac{4u}{v_{IN}} - 4\left(\frac{u}{v_{IN}}\right)^2\right) \frac{u}{v_{IN}}$$

$$B2 = 4 \left[\left(\frac{u}{v_{IN}}\right)^2 - \left(\frac{u}{v_{IN}}\right)^3 \right] = 4 (\alpha^2 - \alpha^3)$$

$$\text{Maximum } \left| \frac{\partial \eta}{\partial \frac{u}{v_{IN}}} \right| = 0 \quad 8\alpha - 12\alpha^2 = 0$$

$$8 = 12\alpha \quad \alpha = \frac{2}{3} \rightarrow u = \frac{2}{3} v_{IN}$$

$$\rightarrow \eta = 4 \left(\left(\frac{2}{3}\right)^2 - \left(\frac{2}{3}\right)^3 \right) = \frac{16}{27}$$

Geo-Energy Question - Answers

Below you find 10 statements; some are true, some are false.

Please indicate for each statement whether it is true or false.

- | | | |
|--|----|-------|
| The concept of 'accommodation space' in stratigraphy is determined by climate | 1 | false |
| When a sourcerock is buried, it will first produce gas and then, at greater depth, oil will be generated | 2 | false |
| The compressibility factor Z is independent on the type of gas | 3 | false |
| Gypsum is the first mineral which precipitates when a sea or salt lake dries out through evaporation | 4 | true |
| Fluvial point bars have a limited lateral continuity | 5 | true |
| Subsidence and sea-level rise have an opposite effect on marine sedimentation | 6 | false |
| A compressional subsurface stress field generates reverse faulting | 7 | true |
| Reflection seismic data yield higher resolution subsurface images than gravity data | 8 | true |
| Gas hydrates represent the largest fossil fuel resources on the planet | 9 | true |
| I. One barrel-of-oil-equivalent (boe) is approximately equal to 1 m ³ of gas at standard conditions | 10 | false |